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Specific heat and neutron diffraction study on quaternary sulfides BaNd₂CoS₅ and BaNd₂ZnS₅

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Abstract

Magnetic and electrical properties are investigated for quaternary neodymium sulfides $BaNd_2TS_5$ (T=Co, Zn) through the specific heat, neutron diffraction, and electrical conductivity measurements. Their electrical conductivities show semiconductive behavior, which follows the Arrhenius temperature dependence with the activation energy of $E_a = 1.46 \text{ eV}$ for $BaNd_2ZnS_5$ and $E_a = 1.19 \text{ eV}$ for $BaNd_2CoS_5$. The specific heat of $BaNd_2ZnS_5$ has a λ -type anomaly at 2.8 K due to the antiferromagnetic ordering of the Nd³⁺ moments and a Schottky-type anomaly at around 60 K, which results from the crystal field splitting of the ⁴ $I_{9/2}$ ground state of the Nd³⁺ ion. The specific heat of $BaNd_2CoS_5$ shows two λ -type anomalies at 5.7 K due to the antiferromagnetic ordering of Nd³⁺ and at 58 K due to the antiferromagnetic ordering of Co²⁺. The latter overlaps with the Schottky-type anomaly due to the crystal field splitting of the Nd³⁺ ion. Neutron diffraction measurements for $BaNd_2CoS_5$ show that a magnetic arrangement of the Co²⁺ moments has a collinear antiferromagnetic structure, while that of the Nd³⁺ moments has a noncollinear one. (C) 2003 Elsevier Science (USA). All rights reserved.

Keywords: Specific heat; Neutron diffraction; Quaternary sulfide; Antiferromagnetic transition; Magnetic structure

1. Introduction

Many binary and ternary compounds containing rare earth and chalcogen have been synthesized [1,2]. These compounds adopt a wide range of structure types and show a variety of interesting physical properties [3]. However, studies on quaternary rare-earth chalcogenides are still scarce, compared with the case for binary and ternary compounds [4–9].

A series of quaternary transition metal sulfides $BaLn_2TS_5$ (Ln = lanthanides, T = Mn, Fe, Co, Zn) crystallizes in a tetragonal structure with the space group, I4/mcm [10–13]. In these sulfides, the transition metal ions are bonded to four sulfur ions in a distorted tetrahedral coordination form and these TS_4 tetrahedra link via the Ba ions. The Ln ion is bonded to eight sulfur neighbors with the D_{2h} site symmetry [11]. Fig. 1 shows the schematic structure of $BaLn_2TS_5$.

Through the magnetic susceptibility measurements of $BaLn_2TS_5$, the Mn, Co, and Fe ions were found to be in the divalent state and were observed to occur an antiferromagnetic transition at $\sim 60 \,\mathrm{K}$ for Mn, at ~60 K for Co, and at ~40 K for Fe [11,12,14]. The powder neutron diffraction data of $BaLa_2TS_5$ (T = Mnand Co) indicated that the Mn^{2+} and Co^{2+} moments had a collinear antiferromagnetic structure at 10K [15,16]. In the paramagnetic temperature region, the effective magnetic moments of the Ln ions in the $BaLn_2TS_5$ were in good agreement with the theoretical moments of the free Ln^{3+} ions. In addition, with decreasing temperature the Nd³⁺ ions also showed an antiferromagnetic ordering below 3-6K in the Nd compounds [12,14]. It seems that the magnetic moments of the Nd^{3+} ions and T^{2+} ions order independently below each of the antiferromagnetic transition temperatures. This magnetic behavior is understandable from its crystal structure, i.e., the BaTS₄ layers and LnS layers are perpendicular to the *c*-axis and they are stacked alternately.

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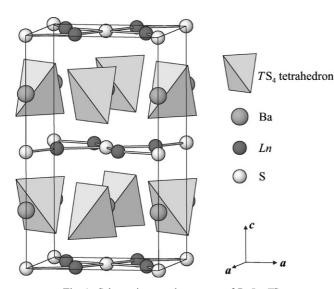


Fig. 1. Schematic crystal structure of BaLn₂TS₅.

In this study, we have paid our attention on two quaternary neodymium sulfides $BaNd_2MS_5$ (T=Co, Zn). Since the magnetic moments of two Nd ions greatly contribute to the magnetic susceptibilities of $BaNd_2CoS_5$, the antiferromagnetic ordering of Co^{2+} ions were not clear in its magnetic susceptibilities vs temperature curve. So, we have determined the Néel temperature of the Co^{2+} ions by calculating the first derivatives of the susceptibility against temperature [12].

In order to obtain additional experimental information and quantitative data on the magnetic ordering of Co^{2+} and Nd^{3+} ions, we have performed specific heat and neutron diffraction measurements on $\text{BaNd}_2\text{CoS}_5$ down to 1.8 K. Specific heat measurements have also been performed on $\text{BaNd}_2\text{ZnS}_5$ to estimate the magnetic entropy change due to only the antiferromagnetic ordering of the Nd^{3+} ions. Furthermore, the electrical properties of both compounds have been investigated through electrical conductivity measurements.

2. Experimental

Quaternary neodymium sulfides $BaNd_2CoS_5$ and $BaNd_2ZnS_5$ were prepared by a solid-state reaction as described elsewhere [12]. The electrical conductivity of the sintered sample was measured by the DC four-probe method in the temperature range from 260 to 400 K. The specific heat was measured by a relaxation technique using a physical property measurement system (Quantum Design, PPMS) in the temperature range from 1.8 to 300 K. The sample in the form of pellet (~10 mg) was mounted on an aluminum plate with apiezon for better thermal contact.

Powder neutron diffraction measurements were carried out using the Kinken powder diffractometer for high-efficiency and high-resolution measurements, HERMES, of Institute for Material Research, Tohoku University, installed at the JRR-3M reactor in Japan Atomic Energy Research Institute, Tokai. Neutrons with wavelength of 1.8196 Å were obtained by the 331 reflection of the Ge monochromator and 12'-blank-22' collimation [17]. The sample was set in a vanadium cylinder with a diameter of 10 mm and sealed in a standard aluminum cell with helium gas and was cooled down to low temperatures using a liquid helium cryostat. Intensity data from 3° to 140° were used in the crystal structure and magnetic structure refinements using Rietveld method program FullProf [18].

3. Results and discussion

3.1. Electrical conductivities

Fig. 2 shows the electrical conductivities (σ) of BaNd₂ZnS₅ and BaNd₂CoS₅ as a function of reciprocal temperature. Both the conductivities increase with temperature and follow the Arrhenius temperature dependence, $\sigma \propto \exp(-E_a/k_BT)$ with the activation energy of E_a . The activation energies were estimated to be 1.46 eV for BaNd₂ZnS₅ and 1.19 eV for BaNd₂CoS₅. The semiconductive behavior of these compounds indicate that the 4*f* and 3*d* electrons are localized on the Nd and Co ions, respectively, which is consistent with the result of their magnetic susceptibility measurements [12].

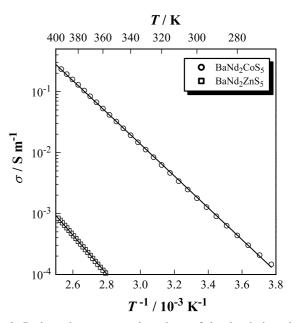


Fig. 2. Reciprocal temperature dependence of the electrical conductivity σ of BaNd₂*T*S₅ (*T*=Co, Zn). Straight solid lines present the Arrhenius law fittings.

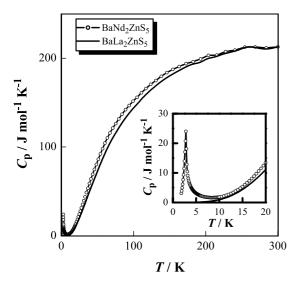


Fig. 3. Temperature dependence of the specific heat C_p of BaNd₂ZnS₅ and BaLa₂ZnS₅ below 300 K. The inset shows C_p below 20 K.

3.2. Specific heats

Fig. 3 shows the temperature dependence of the molar specific heats (C_p) of BaNd₂ZnS₅. A λ -type anomaly, which is attributable to an antiferromagnetic ordering of the Nd^{3+} ions, is observed at 2.8 K as shown in the inset of Fig. 3. This experimental result is consistent with our magnetic susceptibility measurements [12]. The temperature dependence of the specific heats of a nonmagnetic compound, BaLa₂ZnS₅, is also shown in Fig. 3. If we assume that the electronic and lattice contributions to the total C_p are equal between BaNd₂ZnS₅ and BaLa₂ZnS₅, the magnetic specific heat (C_{mag}) of the Nd^{3+} ions is obtained by subtracting the specific heat of $BaLa_2ZnS_5$ from that of $BaNd_2ZnS_5$. The temperature dependences of C_{mag} and the magnetic entropy $S_{\text{mag}} =$ $\int_0^1 (C_{\text{mag}}/T) \, \mathrm{d}T$ are shown in Fig. 4(a) and (b), respectively.

According to the antiferromagnetic spin-wave model, C_{mag} for antiferromagnetic magnons is expected to be proportional to T^3 at $T < T_N$ (~2.8 K) [19]. The calculated values of C_{mag} below 1.8 K are plotted as a broken line in Fig. 4(a). In the $C_{\text{mag}} - T$ curve, another broad peak appears at around 60 K except for the λ -type peak at 2.8 K. The ${}^4I_{9/2}$ ground state of the Nd³⁺ ion should split into five Kramers' doublets by the crystalline electric field in the site symmetry of D_{2h} . In the antiferromagnetic state, only the lowest doublet needs to be considered, because

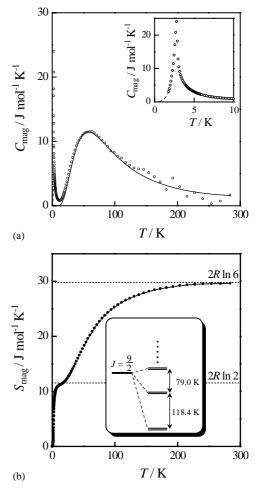


Fig. 4. (a) Temperature dependence of the magnetic specific heat C_{mag} of BaNd₂ZnS₅. The inset shows C_{mag} below 10 K. A broken line indicates the calculated C_{mag} below T_{N} (see text). The fitted curve (solid line) is the Schottky curve from the Nd energy levels (see text). (b) Temperature dependence of the magnetic entropy S_{mag} of BaNd₂ZnS₅. The inset shows the crystal-field-splitting levels of the Nd³⁺ ion in BaNd₂ZnS₅.

 S_{mag} converges at $2R \ln 2$ ($R \ln 2$ per mole of Nd³⁺ ion) below 10 K, where R is the gas constant. In addition, excited doublets contribute to the Schottky-type anomaly at around 60 K. Considering the energy levels with the energy $\varepsilon_0, \varepsilon_1, \ldots, \varepsilon_m$ of the degeneration g_0, g_1, \ldots, g_m , the partition function (Z) in this system is presented by Z = $\sum_{i=0}^{m} g_i \exp(-\varepsilon_i/kT)$ and the Schottky-type specific heat (C_S) is given by $C_s = \partial/\partial T (RT^2 \partial \ln Z/\partial T)$. Therefore, for the Nd³⁺ ion in BaNd₂ZnS₅, C_S by four excited doublets is calculated by the following equation:

$$C_{S} = 2 \times \frac{\left[\sum_{i=1}^{4} \varepsilon_{i}^{2} \exp(-\varepsilon_{i}/kT)\right] \times \left[1 + \sum_{i=1}^{4} \exp(-\varepsilon_{i}/kT)\right] - \left[\sum_{i=1}^{4} \varepsilon_{i} \exp(-\varepsilon_{i}/kT)\right]^{2}}{kT^{2} \left[1 + \sum_{i=1}^{4} \exp(-\varepsilon_{i}/kT)\right]^{2}},$$
(1)

where ε_i is the energy difference between the ground doublet and the *i*th excited doublet. By fitting Eq. (1) to the $C_{\text{mag}} - T$ curve as a solid line in Fig. 4(a), the first and second excited doublets were found to contribute mainly to the C_{S} anomaly at around 60 K, and the energy differences from the ground-state doublet were estimated to be 118.4(4) K for the first excited doublet and 197.4(8) K for the second excited doublet. The value of $S_{\text{mag}} \sim 2R \ln 6$ at 300 K also indicates that the other two excited doublets scarcely affect the C_{S} behavior at low temperatures.

Fig. 5(a) shows the temperature dependence of the molar C_p of BaNd₂CoS₅. Two λ -type anomalies have been observed at corresponding temperatures with an antiferromagnetic ordering of the Co²⁺ ions (~60 K) and that of Nd³⁺ ions (~6K) for their magnetic susceptibilities [3]. The temperature dependence of C_{mag} for BaNd₂CoS₅ is obtained by subtracting the C_p data of BaLa₂ZnS₅ from those of BaNd₂CoS₅ and is plotted in Fig. 5(b). The temperature dependence of S_{mag} obtained by integrating the C_{mag}/T data is also plotted in Fig. 5(b). For BaNd₂CoS₅, the coordination of the Nd³⁺ ions which are bonded to eight sulfurs, was found

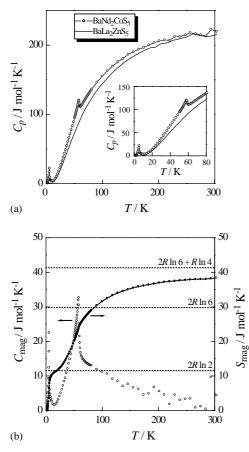


Fig. 5. (a) Temperature dependence of the specific heat C_p of BaNd₂CoS₅ and BaLa₂ZnS₅ below 300 K. The inset shows C_p below 80 K. (b) Temperature dependence of the magnetic specific heat C_{mag} and the magnetic entropy S_{mag} of BaNd₂CoS₅.

to be nearly the same as that for BaNd₂ZnS₅ through the powder X-ray diffraction measurements [12]. Thus, the crystal-field-splitting energy levels of the ${}^{4}I_{9/2}$ ground state of the Nd^{3+} ions in BaNd₂CoS₅ are expected to be similar to those in BaNd₂ZnS₅. For the Nd³⁺ ions in BaNd₂CoS₅, it is considered that the ground-state doublet constitutes the antiferromagnetic state and the first and second excited Kramers' doublets yield the Schottky anomaly at around 60 K in the $C_{\text{mag}} - T$ curve from the analogy with the temperature dependence of C_{mag} and S_{mag} for BaNd₂ZnS₅. However, it is difficult to estimate the energy levels for the first and second excited doublets, because the Schottky anomaly at around 60 K overlaps with the λ -type anomaly due to the antiferromagnetic ordering of the Co^{2+} ions. The value of $S_{\text{mag}} = 38.2 \text{ J mol}^{-1} \text{ K}^{-1}$ at 300 K is slightly smaller than $R (2 \ln 6 + \ln 4)$, where $2R \ln 6$ and $R \ln 4 (= R \ln (2S + 1))$ are attributable to three doublets of the Nd³⁺ ions and the spin state $S = \frac{3}{2}$ of the Co²⁺ ion, respectively.

3.3. Crystal and magnetic structures of BaNd₂CoS₅

Fig. 6 shows the powder neutron diffraction patterns of $BaNd_2CoS_5$ at 2.3, 10, and 70 K. The Co and Nd ions showed an antiferromagnetic ordering below 58 K

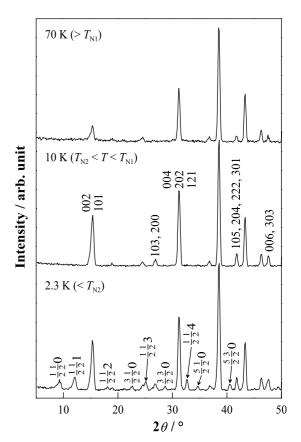


Fig. 6. Powder neutron diffraction patterns for $BaNd_2CoS_5$ at 2.3, 10, and 70 K.

($\equiv T_{\rm N1}$) and 5.7 K ($\equiv T_{\rm N2}$), respectively, through the magnetic susceptibility [12] and the specific heat measurements. Thus, the diffraction pattern at 70 K ($> T_{\rm N1}$) can be indexed on the basis of the chemical cell. The data at 70 K indicate that BaNd₂CoS₅ crystallizes in a tetragonal structure with space group of *I4/mcm* and no structural phase transition occurs between 70 K and room temperature. Table 1 lists the refined lattice parameters and atomic positions of BaNd₂CoS₅ at 70 K. The lattice parameters at 70 K refined by the Rietveld analysis are as follows: a = 7.7816(1) Å, c = 13.5189(4) Å (reliability factors of the refinement: $R_{\rm wp} = 11.0\%$, $R_{\rm Bragg} = 4.01\%$).

Table 1

Crystal and magnetic data determined by neutron diffraction at 70 and 10 K for $BaNd_2CoS_5$

| | Site | x | У | Ζ | $B/{ m \AA}^2$ |
|------|-------------|-----------|--------|-----------|----------------|
| 70 K | | | | | |
| Ba | 4a | 0 | 0 | 1/4 | 0.01(5) |
| Nd | 8h | 0.1624(2) | 0.6624 | 0 | 0.08(3) |
| Со | 4b | 0 | 1/2 | 1/4 | 0.54(13) |
| S(1) | 4c | 0 | 0 | 0 | 0.54(12) |
| S(2) | 16 <i>l</i> | 0.1467(3) | 0.6517 | 0.6342(2) | 0.25(1) |
| 10 K | | | | | |
| Ba | 4a | 0 | 0 | 1/4 | 0.01(6) |
| Nd | 8h | 0.1622(2) | 0.6622 | 0 | 0.07(4) |
| Co | 4b | 0 | 1/2 | 1/4 | 0.26(10) |
| S(1) | 4c | 0 | 0 | 0 | 0.45(11) |
| S(2) | 16 <i>l</i> | 0.1470(3) | 0.6470 | 0.6345(2) | 0.30(7) |
| . / | | | | | |

Space group: I4/mcm

70 K $R_{wp} = 11.0\%$, $R_{Bragg} = 4.01\%$ a = 7.7816(1) Å, c = 13.5189(4) Å 10 K $R_{wp} = 10.6\%$, $R_{Bragg} = 3.60\%$, $R_{mag} = 4.76\%$ a = 7.7782(1) Å, c = 13.5132(4) Å, $m = 3.44(4) \mu_{B}$

Note:
$$R_{wp} = \left[\sum w(y(o) - y(c))^2 / \sum wy(o)^2\right]^{1/2}, R_{Bragg} = \sum |I(o) - I(c)| / \sum wy(o)^2 = \sum |I(o) - I(c)| / \sum |I(o)$$

$$\sum I(o), R_{\text{mag}} = \sum |I_{\text{mag}}(o) - I_{\text{mag}}(c)| / \sum I_{\text{mag}}(o)$$

In the diffraction pattern at 10 K ($T_{N2} < T < T_{N1}$), no additional reflections appear, but magnetic Bragg intensities, which arise from scattering on ordered magnetic moments of the Co ions, are observed for the {*hkl*} reflections with even *h*, *k*, *l* or with odd *h* + *k* and *l* as compared with the pattern at 70 K. The former reflection condition, which satisfies with a ferromagnetic structure, is excluded for the antiferromagnetic ordering of the Co²⁺ moments. The Co ions occupy the 4*b* site with the following coordinates:

Co(1)
$$(\frac{1}{2}, 0, \frac{1}{4})$$
, Co(2) $(\frac{1}{2}, 0, \frac{3}{4})$,
Co(3) $(0, \frac{1}{2}, \frac{1}{4})$, Co(4) $(0, \frac{1}{2}, \frac{3}{4})$.

Then, the collinear antiferromagnetic structure of the Co^{2+} moments has three possible ordering models:

Model I : $m_1 - m_2 + m_3 - m_4$, Model II : $m_1 + m_2 - m_3 - m_4$, Model III : $m_1 - m_2 - m_3 + m_4$,

where m_i is the magnetic moment of the Co(*i*) ion. Only the magnetic ordering of model III satisfies the $\{hkl\}$ reflection condition with odd h + k and l. The Rietveld analysis by using model III determined that the magnetic moment of Co^{2+} was 3.44(4) μ_{B} and oriented along the *c*-axis. The best result of the Rietveld fitting is shown in Fig. 7 and the refined crystallographic and magnetic parameters are summarized in Table 1. The schematic magnetic structure is illustrated in Fig. 8. This magnetic ordering is isostructural with that of BaLa₂₋ CoS_5 and its magnetic moment agrees with the Co^{2+} moment of $3.53 \mu_B$ in BaLa₂CoS₅ at 10 K [16]. The antiferromagnetic transition temperature (58.1 K) of Co^{2+} for BaNd₂CoS₅ was found to be in agreement with that (63.2 K) for BaLa₂CoS₅ through the specific heat measurements [16]. From these results, it is inferred that the magnetic interactions between the Co^{2+} and Nd³⁺ moments are extremely weaker than those between the Co²⁺ moments, and that the Nd³⁺

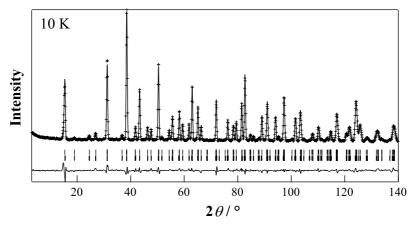


Fig. 7. Powder neutron diffraction pattern and Rietveld refinements for $BaNd_2CoS_5$ at 10 K. The magnitude of magnetic moments is determined to be 3.44(4) μ_B . Reliability factors of the refinement are as follows: $R_{Bragg} = 3.60\%$, $R_{mag} = 4.76\%$.

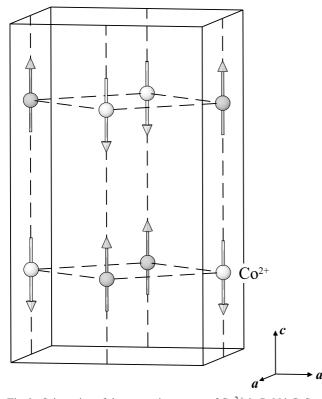


Fig. 8. Orientation of the magnetic moment of Co^{2+} in BaNd₂CoS₅ at 10 K.

moments scarcely affect the magnetic orderings of Co^{2+} .

In the diffraction pattern at 2.3 K ($\langle T_{N2}\rangle$, additional magnetic peaks, which are due to an antiferromagnetic ordering of the Nd³⁺ moments, are observed for the $\{\frac{hk}{22}l\}$ reflections with odd *h* and *k* as compared with the pattern at 10 K. Then, the magnetic unit cell at 2.3 K has a dimension $\sqrt{2a} \times \sqrt{2a} \times c$ in terms of the tetragonal chemical cell. In this unit cell, the Nd³⁺ ions occupy the sites with the following coordinates:

 $\begin{array}{ll} \mathrm{Nd}(1) \; (\frac{1}{4}, \frac{1}{4} + x, 0), & \mathrm{Nd}(2) \; (\frac{1}{4}, \frac{1}{4} - x, 0), \\ \mathrm{Nd}(3) \; (\frac{3}{4} - x, \frac{1}{4}, 0), & \mathrm{Nd}(4) \; (\frac{3}{4} + x, \frac{1}{4}, 0), \\ \mathrm{Nd}(5) \; (\frac{3}{4}, \frac{1}{4} + x, \frac{1}{2}), & \mathrm{Nd}(6) \; (\frac{3}{4}, \frac{1}{4} - x, \frac{1}{2}), \\ \mathrm{Nd}(7) \; (\frac{1}{4} - x, \frac{1}{4}, \frac{1}{2}), & \mathrm{Nd}(8) \; (\frac{1}{4} + x, \frac{1}{4}, \frac{1}{2}), \end{array}$

where x is identical with a positional parameter x in the 8h site, $(x, x - \frac{1}{2}, 0)$, for the chemical cell with the space group, *I*4/*mcm*. In addition, more eight sites Nd(*i*) (*i* = 9, 10, ..., 16) were generated by translation of $(\frac{1}{2}, \frac{1}{2}, 0)$

for Nd(*i*) (i = 1, 2, ..., 8). The magnetic reflection condition for the $\{\frac{hk}{22}l\}$ reflection with odd *h* and *k* leads that the magnetic moment of the Nd(*i*) ion is antiparallel to that of the Nd(*i*+8) ion. From the magnetic Bragg intensities of the Nd³⁺ moments, it was found that the Nd³⁺ magnetic structure was not a collinear antiferromagnetic structure. Unfortunately, we could not determine the magnetic structure of the Nd³⁺ sublattice, because there are too many sites for the Nd³⁺ ions, 16 sites per magnetic unit cell.

References

- J. Flahaut, P. Laruelle, in: L. Eyring (Ed.), Progress in the Science and Technology of the Rare Earths, Vol. 3, Pergamon Press, Oxford, 1968, pp. 149–208.
- [2] J. Flahaut, P. Laruelle, in: L. Eyring (Ed.), Progress in the Science and Technology of the Rare Earths, Vol. 3, Pergamon Press, Oxford, 1968, pp. 209–283.
- [3] J. Flahaut, in: K.A. Gschneidner Jr., L.R. Eyring (Eds.), Handbook on the Physics and Chemistry of Rare Earths, Vol. 4, North-Holland Publishing Company, Amsterdam, New York, Oxford, 1979, pp. 1–88.
- [4] P.M. van Calcar, P.K. Dorhout, Mater. Sci. Forum 315–317 (1999) 322–330.
- [5] W. Ping, J.A. Ibers, J. Solid State Chem. 107 (1993) 347-355.
- [6] W. Ping, A.E. Christuk, J.A. Ibers, J. Solid State Chem. 110 (1994) 337–344.
- [7] W. Ping, J.A. Ibers, J. Solid State Chem. 110 (1994) 156-161.
- [8] R.L. Gitzendanner, C.M. Spencer, F.J. DiSalvo, M.A. Pell, J.A. Ibers, J. Solid State Chem. 131 (1997) 399–404.
- [9] P. Stoll, P. Durichen, C. Nather, W. Bensch, Z. Anorg. Allg. Chem. 624 (1998) 1807–1810.
- [10] H. Masuda, T. Fujino, N. Sato, K. Yamada, J. Solid State Chem. 146 (1999) 336–343.
- [11] M. Wakeshima, Y. Hinatsu, J. Solid State Chem. 153 (2000) 330–335.
- [12] M. Wakeshima, Y. Hinatsu, J. Solid State Chem. 159 (2001) 163–169.
- [13] K. Ino, M. Wakeshima, Y. Hinatsu, Mater. Res. Bull. 36 (2001) 2207–2213.
- [14] M. Wakeshima, K. Ino, Y. Hinatsu, Y. Ishii, Bull. Chem. Soc. Jpn. (2003), in press.
- [15] M. Wakeshima, Y. Hinatsu, K. Oikawa, Y. Shimojo, Y. Morii, J. Mater. Chem. 10 (2000) 2183–2185.
- [16] M. Wakeshima, Y. Hinatsu, Y. Ishii, Y. Shimojo, Y. Morii, J. Mater. Chem. 12 (2002) 631–634.
- [17] K. Ohoyama, T. Kanouchi, K. Nemoto, M. Ohashi, T. Kajitani, Y. Yamaguchi, Jpn. J. Appl. Phys. 37 (1998) 3319–3326.
- [18] J. Rodrigues-Carvajal, Physica B 192 (1993) 55-69.
- [19] S.J. Joshua, A.P. Cracknell, Phys. Lett. A 28 (1969) 562-563.